Ligand-accelerated enantioselective methylene C(sp³)-H bond activation

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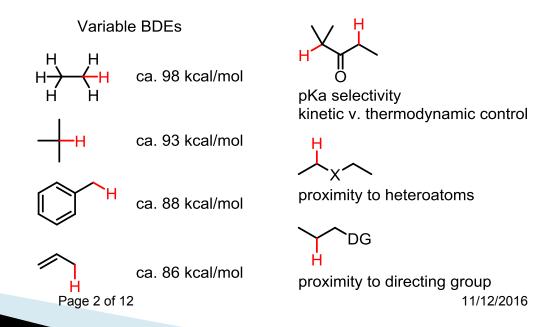
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Science 2016, 353, 1023.

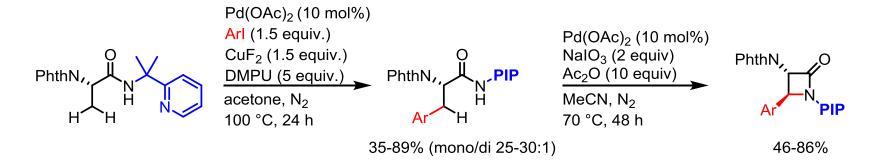
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Current Literature 09/10/16
Wipf Group

C(sp³)-H Functionalization

- Challenges:
 - Reaction efficiency
 - Functional group compatibility/reaction scope
 - Regio- and stereo-selectivity
 - Atom economy (directing groups are very large and add linear steps)
- Not all C(sp³)-H are created equally!



C(sp³)-H Functionalization by the Shi Group



Angew. Chem. Int. Ed. 2013, 52, 13588.

Pd(OAc)₂ (10 mol%)

$$K_2CO_3$$
 (2.5 equiv)
PivOH (0.2 equiv)
 t -BuOH, N_2
120 °C, 24 h

Chem. Commun. 2014, 50, 8353.

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C(sp³)-H Functionalization by the Yu Group

H O F CF3

92%, mono:di = 1:8

JACS 2012, 134, 18570.

$$\begin{array}{c|c}
 & F \\
 & F \\
 & F \\
 & F
\end{array}$$

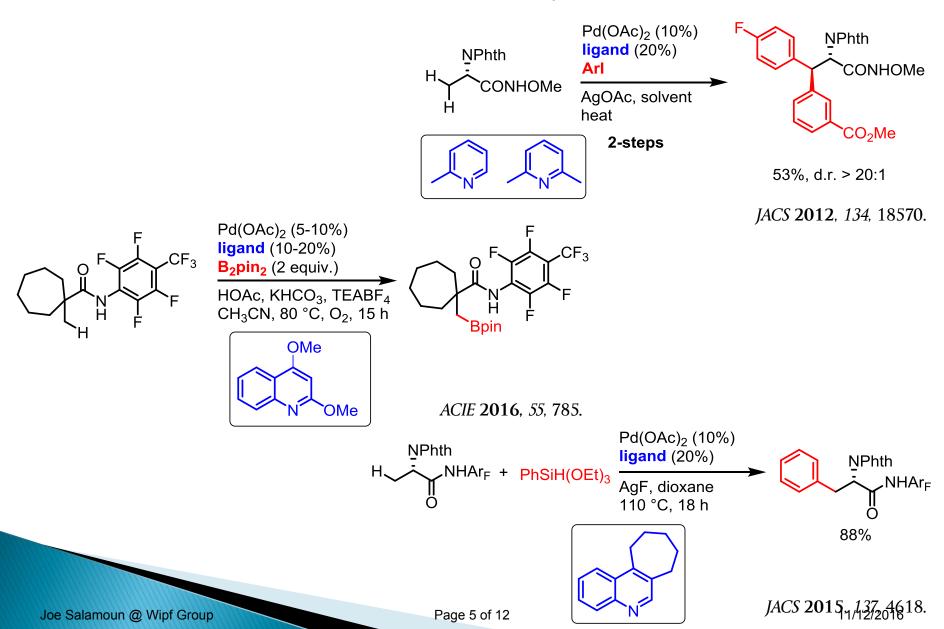
$$\begin{array}{c}
 & CF_3 \\
 & F \\
 & OBz
\end{array}$$

71%

Science 2016, 351, 252.

ACIE **2015**, *54*, 6545. 11/12/2016

C(sp³)-H Functionalization by the Yu Group



Ligand Optimization

Starting Material and Ligand Synthesis

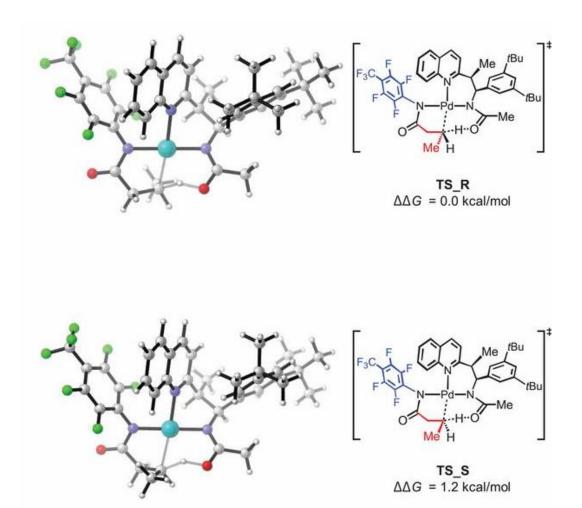
1.
$$(COCI)_2$$
, CH_2CI_2
2. Ar_FNH_2 , toluene reflux

1. n -BuLi, THF
0 °C to rt, 3 h

1. t -Bu
1. t -Bu
1. t -Bu
4 isomers, 75%
(S,S) = 23%

1. t -Bu

DFT Derived Transition State Structures



Aryl Iodide Scope

2t, 61%, 94:6 er

2s, 60%, 93.5:6.5 er

2v, 56%, 92:8 er

2u, 58%, 94:6 er

Scope of Amides and Acids

R = Et**4a**, 78%, 93.5:6.5 er R = -(CH₂)₂Ph**4k**, 72%, 90:10 er n-Pr 4b, 83%, 95:5 er -(CH₂)₄CO₂Me **4I**, 60%, 95:5 er 4c, 80%, 94:6 er 4m, 68%, 95:5 er *n*-amyl -(CH₂)₅CO₂Me $-(CH_2)_{10}Me$ 4d, 79%, 94:6 er **4n**, 50%, 93.5:6.5 er -(CH₂)₃NPhth **4e**, 48%, 89.5:10.5 er -(CH₂)₃OMe **4o**, 40%, 95.5:4.5 er cyclopentyl **4f**, 56%, 93.5:6.5 er cyclohexyl 4g, 71%, 95:5 er 4j, 81%, 95:5 er **4h**, 72%, 94:6 er 4i, 74%, 95:5 er TsN 4p, 35%, 94:6 er **4q**, 68%, 95:5 er 4r, 37%, 96:4 er **4s**, 57%, 93:7 er

 $NHAr_F R = H$

4-CO₂Me

3-Mel

4t, 85%, 87:13 er **4u**, 87%, 92.5:7.5 er 4v, 85%, 90:10 er

NPhth

4w, 56%, 92:8 er

4x, 42%,

Science **2016**, 353, 1023.

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Poor Atom/Step Economy with DG Chemistry Remains a Major Challenge

$$F_{3}C$$

$$F_{3}C$$

$$F_{3}C$$

$$F_{3}C$$

$$F_{3}C$$

$$F_{4}C$$

$$F_{5}C$$

Combined MW: 539.2323

MW: 178.2310

Combined MW: 395.3410

MW: 178.2310

Science 2016. 353. 1023.

Chem. Commun. 2014, 50, 8353.

Conclusions

Highlights:

- Useful methodology when conjugate addition is difficult.
- Excellent enantioselectivity.
- Ligands are accessible and tunable.

Room for Improvement:

- Are non-aromatic pi bonds (alkenes, alkynes) tolerated?
- ... and heterocycles?
- Reaction times are long and catalyst loading is higher than typical cross-couplings.
- DG accounts for significant loss of atom economy.

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